

Production of molecule on a pyrex surface under plasma exposure: example of NO

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Extensive experimental and theoretical study of NO_x production in low pressure N₂/O₂ containing plasmas in continuous [1,2,3] and pulsed mode [4], as well as computational fluid dynamics simulation of space vehicles re-entry reveal an important role of O and N atom surface-catalyzed recombination into NO molecule. However, in all above cases analysis of the relative yield of this reaction is not straightforward and it requires sophisticated numerical modeling. We propose to study NO production on the surface pretreated in different low pressure plasmas. In ref [5], Authors claim, that O and N atoms are chemically adsorbed on the glass surface with bonding energy of about 5 eV. It allows us to expect, that atoms remain stucked on the discharge tube wall for a long time after plasma exposure.

Experimental procedure consists of several steps – pretreatment with N₂, O₂, Air and Ar plasmas, pumping and cooling of the tube and then O₂ (or N₂) plasma in which we detect NO production. The discharge tube is 2 cm inner diameter and 50 cm long, and the gas pressure is around 1 Torr. In order to eliminate plasma contamination by metallic electrodes inside the reactor, we use CCP discharge symmetrically driven by 13.56 MHz pulsed power supply, the coupling electrodes being located outside the plasma tube. During the plasma phase, the monitoring of the formation/loss of NO is made by in situ time resolved TDLAS. Depending upon the experimental condition, a time resolution as high as 5 ms may be achieved. Experiments may be carried out with a closed reactor (no gas flow) or under flowing conditions.

In N₂ plasma with small admixture of NO, NO molecules are efficiently destroyed in collisions with N-atoms [6], what makes it difficult to detect nitric oxide produced on the surface. On the contrary, after a nitrogen pre-treatment of the pyrex surface, and under O₂ plasma, NO production is evidenced. Figure 1 shows such an example.

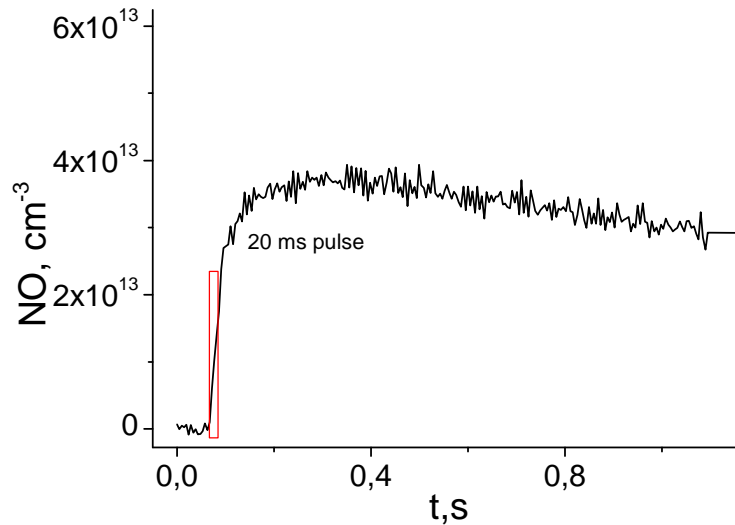


Figure 1. Production of NO molecules in a pure O₂ plasma, after the pyrex surface of the reactor has been treated by 30 mn pure N₂ plasma. The O₂ plasma pulse is 20 ms (single shot), in a closed reactor. The monitoring of NO is made by in situ TDLAS with a time resolution of 5 ms.

It is shown that under O₂ plasma pulses after N₂ plasma pretreatment NO concentration reaches 1-10 10¹³ cm⁻³ depending upon the plasma pulse duration. Such a surface production corresponds to about 5-50 10¹² cm⁻² N atoms grafted on the pyrex surface during the N₂ pretreatment. No NO formation is detected after an argon or oxygen plasma pre-treatment or a neutral nitrogen gas glow. By varying the length of N₂ plasma column, proportionality between pretreated surface and amount of NO produced was verified.

- [1] Y.Ionikh et al. Chemical Physics 322(2006) 411-422
- [2] B.F. Gordiets et al. IEEE Trans. Plasma. Sci. 23(1995) 750.
- [3] R. Zijlmans, Thesis, Eindhoven 2008
- [4] Gatilova et al. Plasma Sources Sci. Technol. 16 (2007) S107–S114
- [5] R. Sayos J. Phys. Chem. B, 2005, 109, 14954-14964.
- [6] C.Pintasilgo et al. submitted to J. Phys. D: Appl. Phys.